







Engineering Approaches to Dendrite-Free Lithium Anodes

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Overview

Timeline

- Project start date:
 - October 1, 2016
- Project end date:
 - September 30, 2019
- Percent complete: 17%

Budget

- Total project funding
 - DOE share: \$1,250,000
 - Contractor share: \$0
- Funding received in FY 2016: \$416,667
- Funding for FY 2017: \$416,667

Barriers

- Barriers addressed
 - Limited Cycle life
 - High cost
 - Low Energy Density

Partners

- Interactions/collaborations
 - UPitt (D. K. Achary)
 - Global Pragmatic Materials (A. Manivannan)
 - Kurt J. Lesker Co. (KJL)
 - Complete Solutions

Relevance/Objectives

- Develop highly stable, long cycle life dendrite free Li metal based anodes with specific capacity ≥2000mAh/g and areal capacity ≥10mAh/cm² in Li metal batteries.
- Develop low cost porous metal alloy foams (PF) for use as suitable current collectors for lithium based anodes to prevent orthogonal dendrite growth and stable SEI layer to enable good cyclability.
- Develop multilayer porous foams (MPFs) by coating PF with Lithium-ion Conductor (LIC) materials to prevent orthogonal growth of dendrites to achieve excellent cycle life.
- Develop low cost Li-rich structurally isomorphous alloys (SIA) electrode structures to completely prevent dendrite nucleation and growth during charge/discharge cycling to improve the performance.
- Improve understanding of the fundamental mechanisms and develop mitigation strategies preventing dendrite nucleation and growth of Li metal anode during cycling.

Milestones-FY 2017 and FY-2018

Date	Description	Type	Status
January 2017	Electrode and cell design parameters for achieving the PEV Target: ≥350Wh/kg, ≥750Wh/l and 10 mAh full cell will be determined	Milestone	Complete
January 2017	Identify and synthesize porous foams materials with high electronic conductivity, porosity>60%, stability in electrolyte and high pore volume to nucleate lithium in-pore	Milestone	Complete
January 2017	Identification and synthesis of LIC materials for use as coatings on both PFs and use in composite multilayer anodes (CMAs) with room temperature ionic conductivity (10 ⁻³ -10 ⁻⁴ S/cm)	Milestone	Complete
July 2017	Synthesis of MPFs exhibiting; Specific capacity ≥1000 mAh/g (≥4 mAh/cm²), >400 cycles without cell failure), initial CE: ≥95% with ≤0.05% loss per cycle (LPC).	Milestone	On-going
October 2017	Identification of suitable materials for SIA development.	Milestone	On-going
October 2017	Fabrication, characterization of desired thickness electrode based on 10 mAh cell, testing electrode performance in coin cell configuration	Milestone	On-going
October 2017	Suitable electrodes will be designed, fabricated and tested in a coin cell configurations and compared to desire AOI6 volumetric energy density requirements. Specific capacity ≥1000 mAh/g (≥4 mAh/cm²), >400 cycles without cell failure)	Go/No-Go	On-going
October 2018	Fabrication and characterization of suitable SIA and CMA electrodes: Half-cell tested CMAs tested as per AOI6 requirements with Specific capacity≥1400 mAh/g (≥4 mAh/cm²), >500 cycles, CE LPC ≤0.05%, CE: ≥97%	Go/No Go	On-going

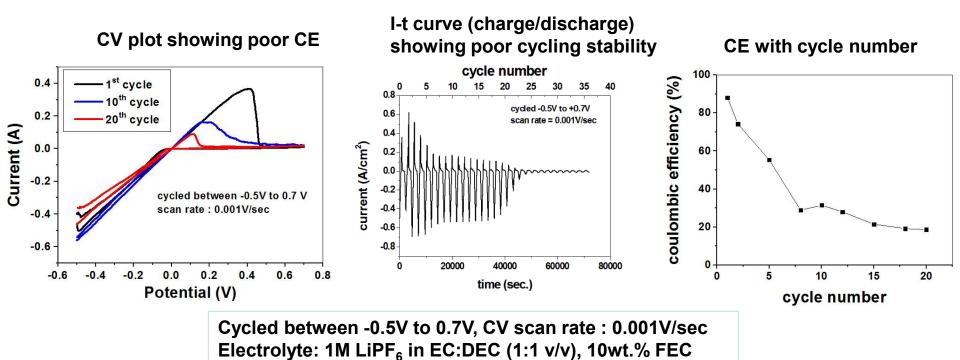
Approach

Several approaches planned to be explored for solving Li-metal dendrite problem:

2016 Oct.	Nov.	Dec.	2017 Jan.	Feb.	Mar.	April	May	June	July	Aug.	Sept.
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Porous foam identification and synthesis				Synthesize LIC coated multilayer porous foams (MPFs)					ng in gress		
				SIA iden	itification			SIA synth			
						Enginee coating				optin	hing and nizing iiques

- ➤ **Go/ No Go decision point:** The Go/ No Go point will be determined October 2017. Suitable progress is being made towards the same.
- ➤ Challenges and barriers: SIA under high rate undergoes surface segregation. Plans in place to optimize composition using first principles calculations.

Fundamental problems of Li metal as an anode: Plating and stripping of Li metal on stainless steel current collector

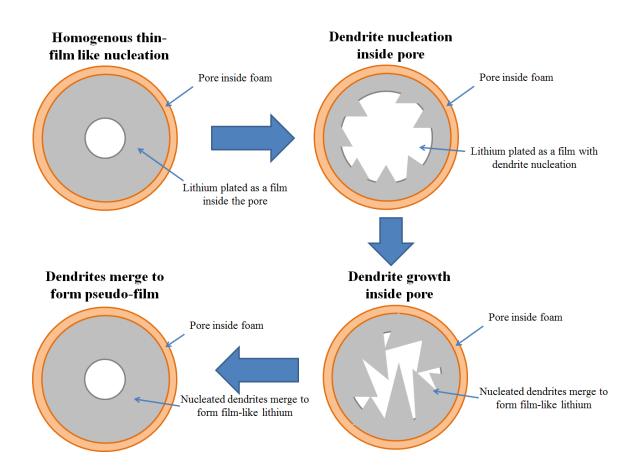


Low CE and poor cycling stability of Li metal anode arises due to nucleation and growth of Li dendrite during cycling which facilities formation of undesired SEI layer in each and every cycle

To enhance the cycle life and formation of stable SEI layer (high CE), an innovative architectural design of current collector or Li metal electrode needs to be engineered to prevent nucleation and growth of Li dendrites.

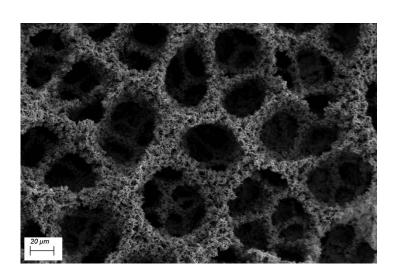
Technical accomplishments: Porous foams (PFs)

- ➤ Use of highly porous electrode architecture coated with a lithium ion conducting insulator (LIC) coating will prevent surface nucleation of lithium and consequently prevent dendrite growth
- In-pore lithium deposition will ensure non-uniform nucleation and growth of lithium leading to coalescence within the pore preventing dendritic growth perpendicular to the separator resulting in puncture and eventual failure due to shorting of cells



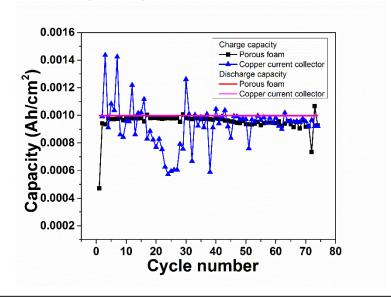
- > Amenable for large electrodes
- > Foam structures result in prevention of orthogonal separator puncture

Innovative architectural design to inhibit the formation of Li-dendrite during cycling: Porous metal alloy foams (PFs) as current collector





Effect of these factors on dendrite formation still under study



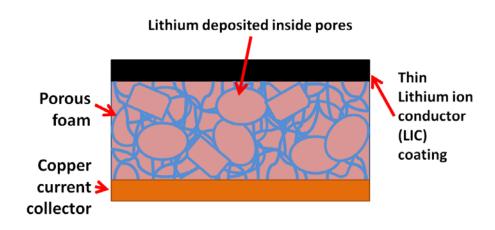
Significant improvement in coulombic efficiency (CE) indicating improved SEI stabilization and limiting dendrite formation with cycling of PF based current collector

✓ In-pore lithium deposition will ensure non-uniform nucleation and growth of lithium leading to coalescence within the pore preventing dendritic growth

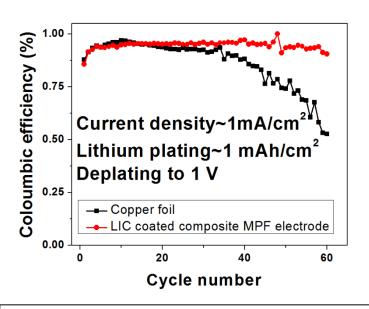
To further enhance the CE and cycle life, and formation of stable SEI layer (high CE), LIC coated PF will be developed

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Innovative architectural design to inhibit the formation of Li-dendrite during cycling: Multilayer Porous foams (MPFs) as current collector



Use of PF, coated with a LIC, is expected to prevent surface nucleation of lithium and consequently prevent dendrite growth

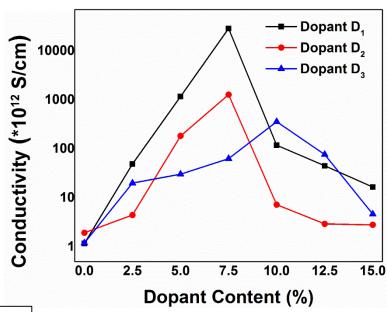


Significant improvement in CE of MPFs indicating improved SEI stabilization and limited dendrite formation with cycling

✓ Engineering design of suitable MPFs with superior Li⁺ conductivity and explore ideal morphologies as a potential solution to ensure dendrite prevention and induce cycling stability

Effect of doping on conductivity of LIC

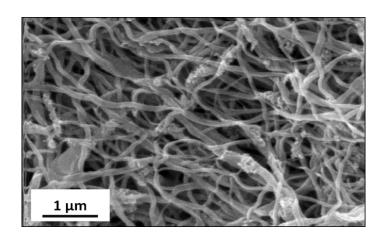
	D ₁ doped (×10 ¹² S/cm)	D ₂ doped (×10 ¹² S/cm)	D ₃ doped (×10 ¹² S/cm)
Li ₄ SiO ₄ - undoped	1.16	1.29	1.19
2.5% doped	49.13	4.44	20.04
5% doped	1169.55	183.51	30.41
7.5% doped	28706.06	1277.59	63.13
10% doped	119.26	7.22	359.62
12.5% doped	45.08	2.93	76.95
15% doped	16.53	2.80	4.66



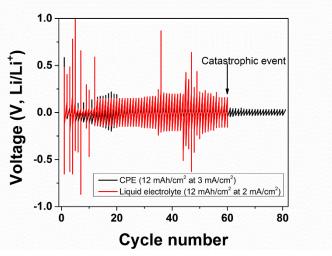
- D₁, D₂ and D₃ doping increases the ionic conductivity of Li₄SiO₄
- 7.5% D₁ doping gives four order increase in conductivity
- 7.5% D₂ doping gives three order increase in conductivity

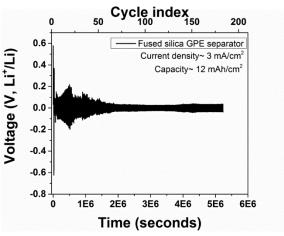
Li Ionic Conductivity as a function of doping content

Innovative architectural design to inhibit formation of Li-dendrites during cycling: Composite Polymeric Electrolyte (CPE) membranes as LICs



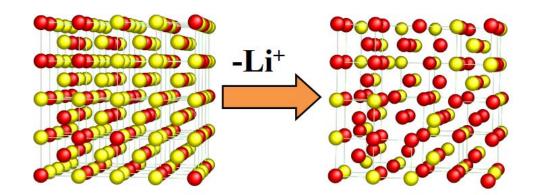
✓ CPE membranes with nanofiller have meshlike morphology aiding in obtaining high mechanical strength resulting in superior electrochemical characteristics





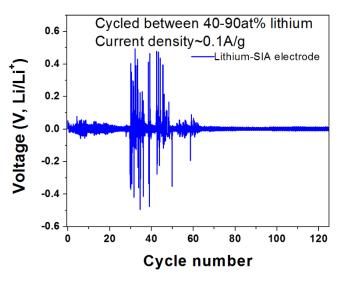
Innovative architectural design to inhibit the formation of Li-dendrite during cycling: Structurally Isomorphous Alloys (SIAs)

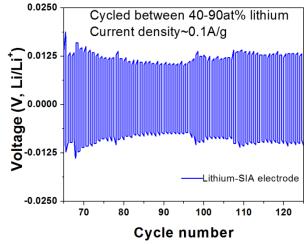
- SIA materials will maintain cubic lithium structure over a wide composition spectrum
- Prevents dendritic growth by adopting a non-nucleation based mechanism



- Theoretical capacity: 1000-2000 mAh/g
- SIA of Li (Li-yellow balls, alloying metal-red balls) which retains its cubic structure even upon removal of ~40% Li
- SIA regime is maintained over ~60 at.% Li

Innovative architectural design to inhibit the formation of Li-dendrite during cycling: Li-SIA electrodes





Working electrode: Lithium-SIA Counter electrode: Lithium metal

electrode

Electrolyte: 1.8M LiCF₃SO₃ in 1, 3

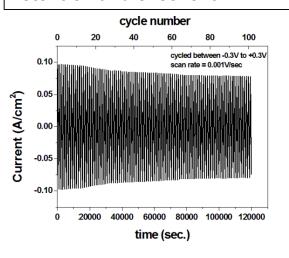
dioxolane and 1, 2

dimethoxyethane (1:1 by vol.) combined with 0.1 M LiNO₃

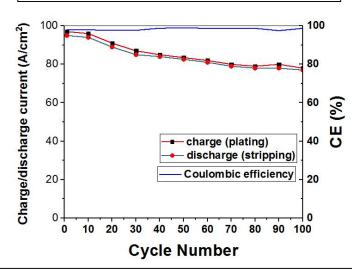
- Initial increase in overpotential due to surface current density
- > Subsequent stable and low overpotential cycling indicative of SEI stabilization and dendrite free cycling
- Use of suitable alloys can improve lithium diffusion preventing surface segregation while continuing to maintain dendrite-free cycling
- Use of SIA cage structures will further decrease solid-state diffusion problems

Innovative architectural design to inhibit the formation of Li-dendrite during cycling: Novel SIA based engineered electrodes

I-t plot showing stable capacity retention and excellent CE



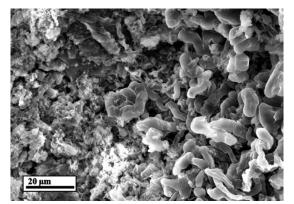
Excellent CE and cycling stability

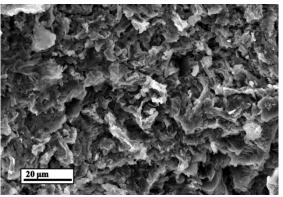


- Cycled between -0.3V to 0.3V.
- CV scan rate : 0.001V/sec.
- Electrolyte: 1M LiPF₆ in EC:DEC (1:1 v/v), 10wt.% FEC

- > Significant improvement of CE using novel engineered SIA electrodes (~98% at 1st cycle to 97% at 100th cycle)
- > Improved cycling stability of Li metal anode in presence of novel SIA based engineered electrodes
- The causes of suppressing the nucleation and growth of Li dendrite will be studied in presence of SIA electrode and the underlying mechanism will be used to engineered perfect SIA electrode with zero capacity fade
 - ✓ Suitable design and engineering of SIA electrodes will help overcome the problem of dendrite formation

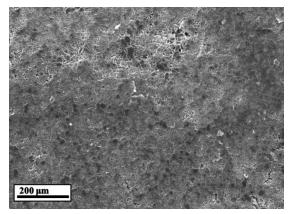
Technical accomplishments: Dendrite-free structures



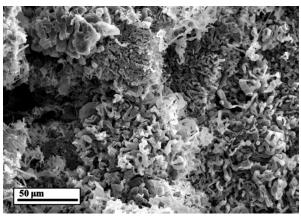


Li counter anode Li-SIA electrode
At 125 cycles with a current rate of
100 mA/g: asymmetric cell

- Lithium electrodes cycled independently and counter electrodes demonstrate dendrite formation
- ➤ Li-SIA electrodes show no formation of lithium dendrites



Li-SIA electrode



Li electrode in a Li-Li symmetric cell showing dendrites

- ➤ Surface shows no dendritic structures
- ➤ Presence of alloying element at the surface confirmed by EDAX analysis

At 125 cycles with a current rate of 100 mA/g: symmetric cell

Responses to Previous Year Reviewers' Comments

 The project was not reviewed last year. Hence there are no comments.

Collaboration and Coordination with Other Institutions

- Collaborators (outside the VT Program):
 - Dr. D. Krishnan Achary (University of Pittsburgh):
 Solid-state nuclear magnetic resonance (MAS-NMR)
 characterization to study the failure mechanisms
 - A. Manivannan (Global Pragmatic Materials):
 Materials characterization using XPS studies
 - Kurt J. Lesker Co. (KJL): Thin film deposition and vacuum techniques
 - Complete Solutions: Technology Translation

Remaining Challenges and Barriers

Challenges

- Overcoming surface phase segregation in SIA electrodes at high rates
- Design of synthesis strategies to make SIA porous foams
- ➤ Interface engineering of multilayer porous foams to ensure good adhesion and strength of LIC and PF interface and LIC-electrolyte interface

> Barriers

- No significant barriers to proposed work has been encountered thus far
- Possible barriers include accounting for scaling of performance and increasing thickness of electrodes

Proposed Future Research

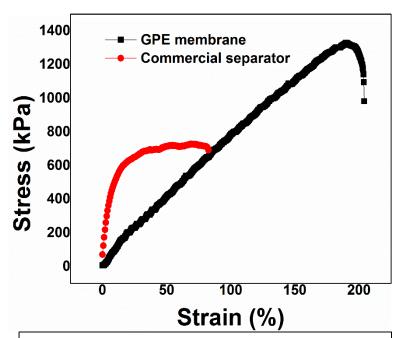
Milestone	Type	Description
Synthesis and testing of SIA electrodes	Technical	Develop SIA materials and electrodes: Specific capacity ≥1400 mAh/g (≥4 mAh/cm²), >500 cycles, CE LPC ≤0.05%, CE: ≥97%
Design and engineering of high capacity CMAs	Technical	Develop LIC coated SIA PFs for use as bulk anode materials by interface/surface coating with Li ⁺ conducting materials. Generate interface engineered CMA architectures exhibiting; Specific capacity ≥1800 mAh/g (≥4 mAh/cm²), ~1000 cycles, CE LPC ≤0.01%, CE: ≥99%.
Optimize MPF electrodes to improve capacity and stability for scaling	Technical	Optimized MPF electrodes with improved capacity and stability
Fabrication and characterization of suitable SIA and CMA electrodes	Go/No Go	Half-cell tested CMAs tested as per AOI6 requirements with Specific capacity≥1400 mAh/g (≥4 mAh/cm²), >500 cycles, CE LPC ≤0.05%, CE: ≥97%

Summary

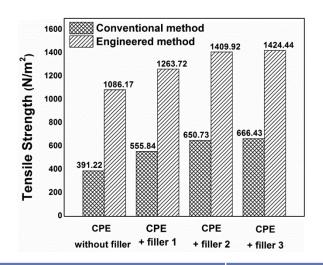
- ➤ Use of multilayer porous foams (MLPF) allows for controlled nucleation of SEI and prevents surface dendrites reflected as improvement in columbic efficiency and cycling stability
- New design of composite polymeric membrane (CPE) materials shown to reduce overpotential in lithium symmetric cells and improve cycle life on account of superior mechanical properties
- ➤ Use of Structurally isomorphous electrodes or current collectors results in high capacity Li electrodes (~1627 mAh/g) with no visible dendrite formation
- SIA electrodes show initial overpotential possibly as a result of surface segregation
- ➤ High rate SIA electrodes demonstrate negligible dendritic structure formation indicating promise of the approach

Technical Back-Up Slides

Innovative architectural design to inhibit formation of Li-dendrite during cycling: Composite Polymeric Electrolyte (CPE) membranes as LICs



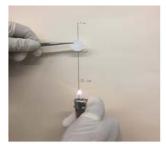
CPEs show superior mechanical stability and tensile strength as compared to commercial separators



Sample Composition	Conductivity (S/cm)
Commercial separator with liquid electrolyte	$1.283 \cdot 10^{-3}$
CPE with filler 1	$1.881 \cdot 10^{-3}$
CPE with filler 2	$3.009 \cdot 10^{-3}$
CPE with filler 3	$9.4749 \cdot 10^{-3}$

Innovative architectural design to inhibit the formation of Li-dendrite during cycling: Composite Polymeric Electrolyte (CPE) membranes as LICs

New Flame resistant CPEs



Commercial separator exposed to heat



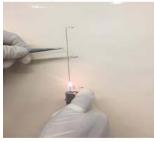
CPE exposed to heat



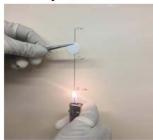
Commercial battery separator



CPE battery separator



Commercial separator shrinking within 5 secs



CPE unaffected by heat for over 60 sec



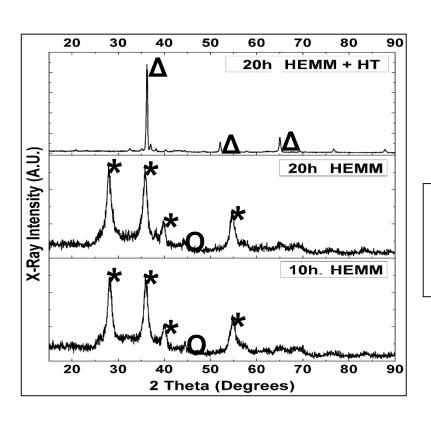
Commercial battery separator upon exposure to fire



CPE battery separator upon exposure to fire

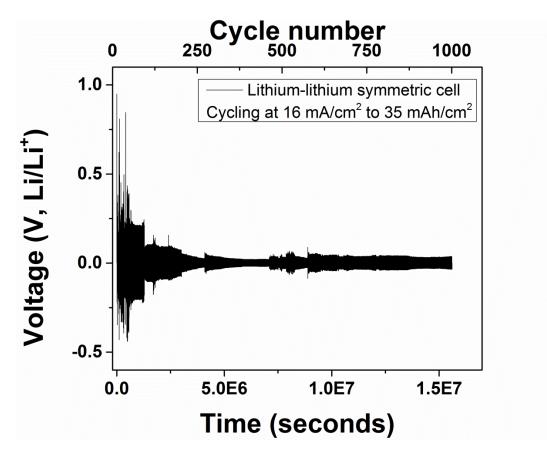
✓ CPEs are intrinsically tolerant to abusive conditions such as short circuits, over charge, over discharge, crush impacts, or exposure to fire and other high-temperature environments

Novel High-throughput Synthesis of Li-SIA using high energy mechanical milling



- Δ-Li-SiA
- * Precursor material
- O Secondary phase
- SIA alloy was synthesized by HEMM followed by heat treatment
- Approach amenable for commercial level scale up

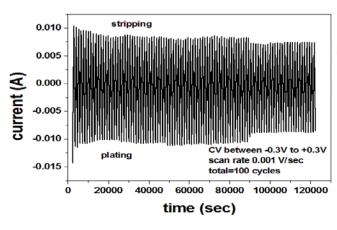
Engineering Cycling Parameters: Potential solution!

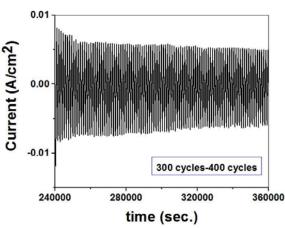


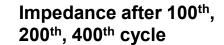
➤ Limited capacity cycling could prevent dendritic pillar formation avoiding separator puncture and consequent shorting

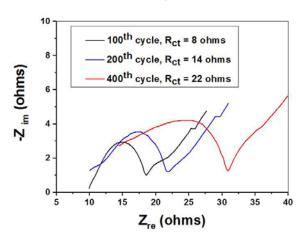
Innovative architectural design to inhibit the formation of Li-dendrite during cycling: LIC coated SIA electrode

I-t curve (charge/discharge) up to 400 cycles showing excellent cycling stability









- Significant improvement of CE and cycling stability of Li metal based anode using LIC coated SIA electrode
- No significant change in R_{ct} with increase in cycle numbers of LIC coated SIA showing promise of novel LIC coated SIA electrode architectures